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Since 1 mol of A and 1 mol of B reacted, it must form 1 mol of P and 1 mol Q - i.e. the

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concentration of products at equilibrium is 1 M. At an earlier time of the reaction, (let's pick a time when $[A] = 0.8 \text{ M}$), only part of the reactants have reacted (in this case 0.2 M), producing an equal amount of products, P and Q. Graphs of $[A]$ and $[P]$ as a function of time are shown below.

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Negatively charged R
groups Aspartate
Glutamate Asp D Glu E
133 147 1.88 2.19 9.60
9.67 3.65 4.25 2.77
3.22 -3.5 —3.5 5.3 6.3

*A scale combining
hydrophobicity and
hydrophilicity of R
groups;

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contribute to the
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